

*54th POWER PLANT COLLOQUIUM
at October 18th and 19th, 2022
International Congress Center Dresden*

**Bromine-Enhanced Mercury Oxidation
at the PGE GiEK Lignite Fired
Power Plant Bełchatów (Poland)**

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Vosteen Consulting GmbH, Köln*

Agenda

1. Introduction
2. History of the BEMO Technology
3. Analysis Prof. Burmistrz at Bełchatów Unit 4 in 2016
4. Hg Emission Control Activities
at Bełchatów Unit 5 and Unit 14 in 2017/ 2018
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8. Conclusions

PGE's Lignite-Fired Power Plants in Belchatów (in total 5,100 MWe):

**All operating Units 2-12 and the new Unit 14 as well
are served with Bromide (Status 2022)**

Unit 14

Units 12 – 7

Units 6 – 2



Widok z lotu ptaka na Elektrownię Bełchatów

380 MWe Units 2 -12 Rafako-Babcock design

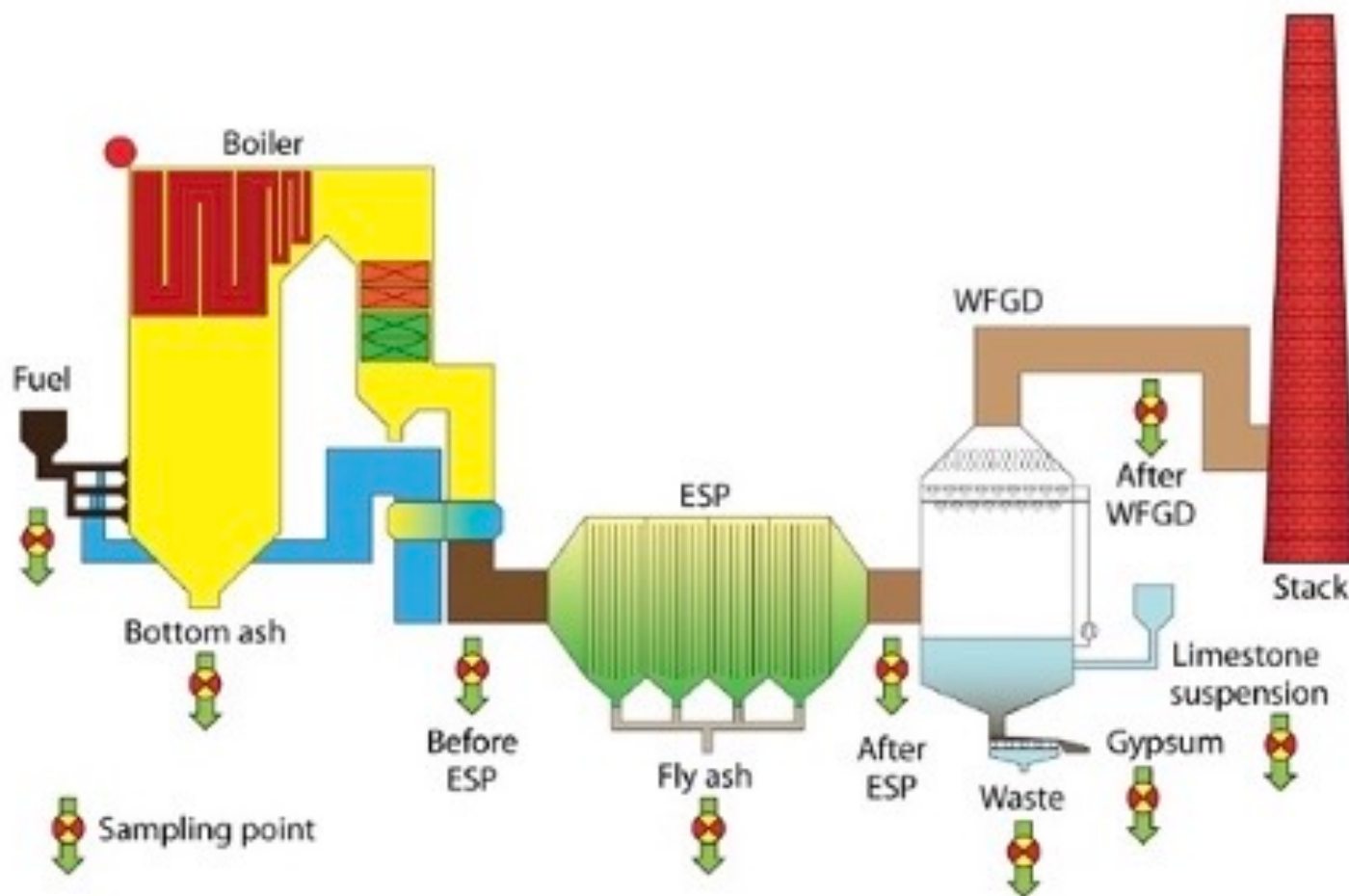
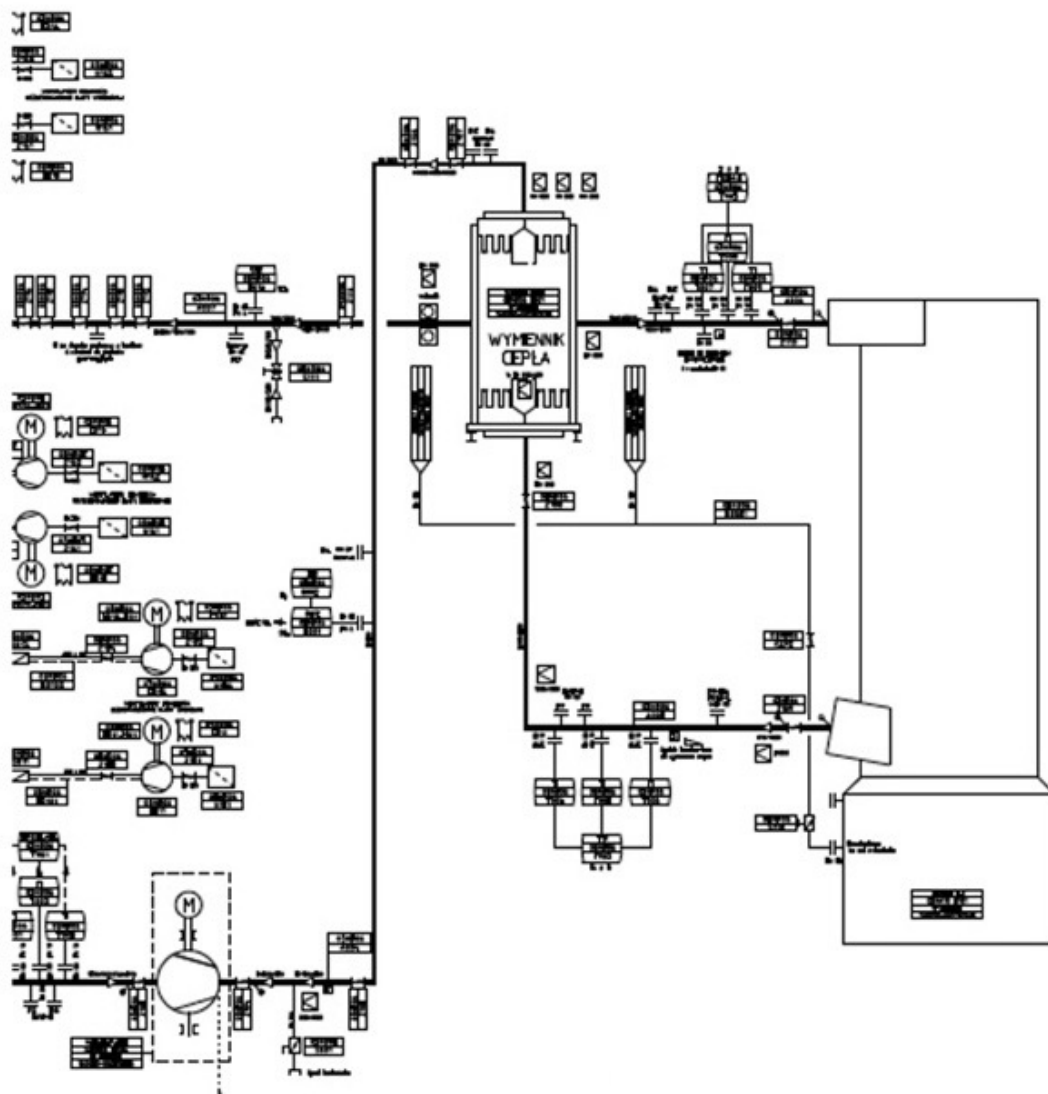


Fig. 2. Schematic diagram of the sampling campaigns in the sampled power plants.

Advantage Bełchatów : There are no rotating REGAVOs



The cross-flow heat exchanger is perfect with respect to complete mercury removal („no bypass of flue gas around the FGD, streaming from the FGD raw gas to the FGD clean gas“)

STEAG-Kraftwerk Ruhr operating with rotating ReGaVo:

Transfer of oxidized mercury Hg^{++}
from ESP-outlet to FGD-exit (clean gas)

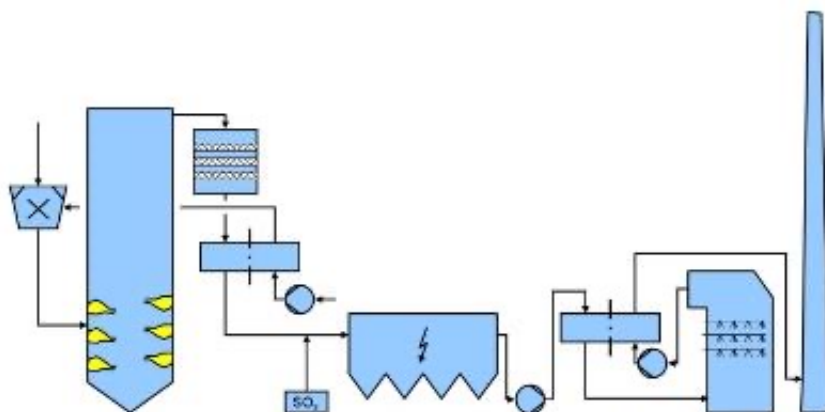


Bild AII-1. Anlagenschema Kraftwerk Ruhr: Rohrkugelmöhlen, Trockenfeuerung (Boxer), High-Dust-DeNOx, LuVo, SO_2 -Konditionierung, E Filter, Saugzug, ReGaVo, REA, Nassgebläse, ReGaVo und trockener Schornstein.

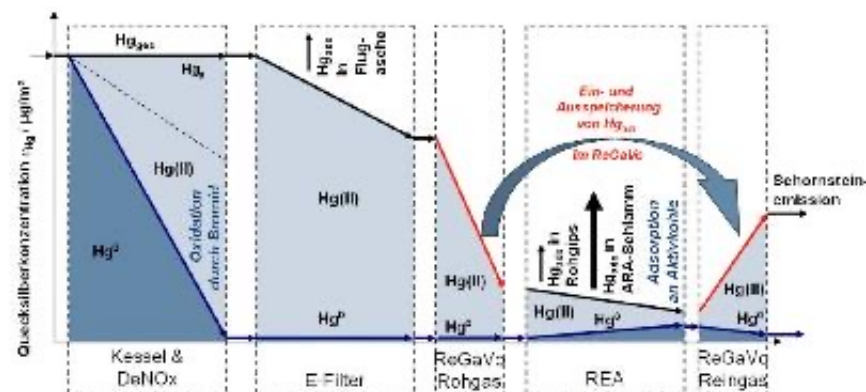


Bild 6-1. Vereinfachte Darstellung des Verlaufs der gasseitigen Hg-Spezifizierung im Kraftwerksprozess des Kraftwerks Ruhr (oder vergleichbar) bei Einsatz der diskutierten Verfahren

Block 14 with 858 MWe, built by Rafako-Alstom



netto el. efficiency of Unit 14: 41,5 %

APC at Unit 14 in Belchatów showing the FGD Lines 1 and 2 (from Google maps)



Continuous Mercury Monitoring Systems

CMM AutoQAL and CMM

Gasmet offers two solutions for Continuous Mercury Monitoring, the CMM and CMM AutoQAL.

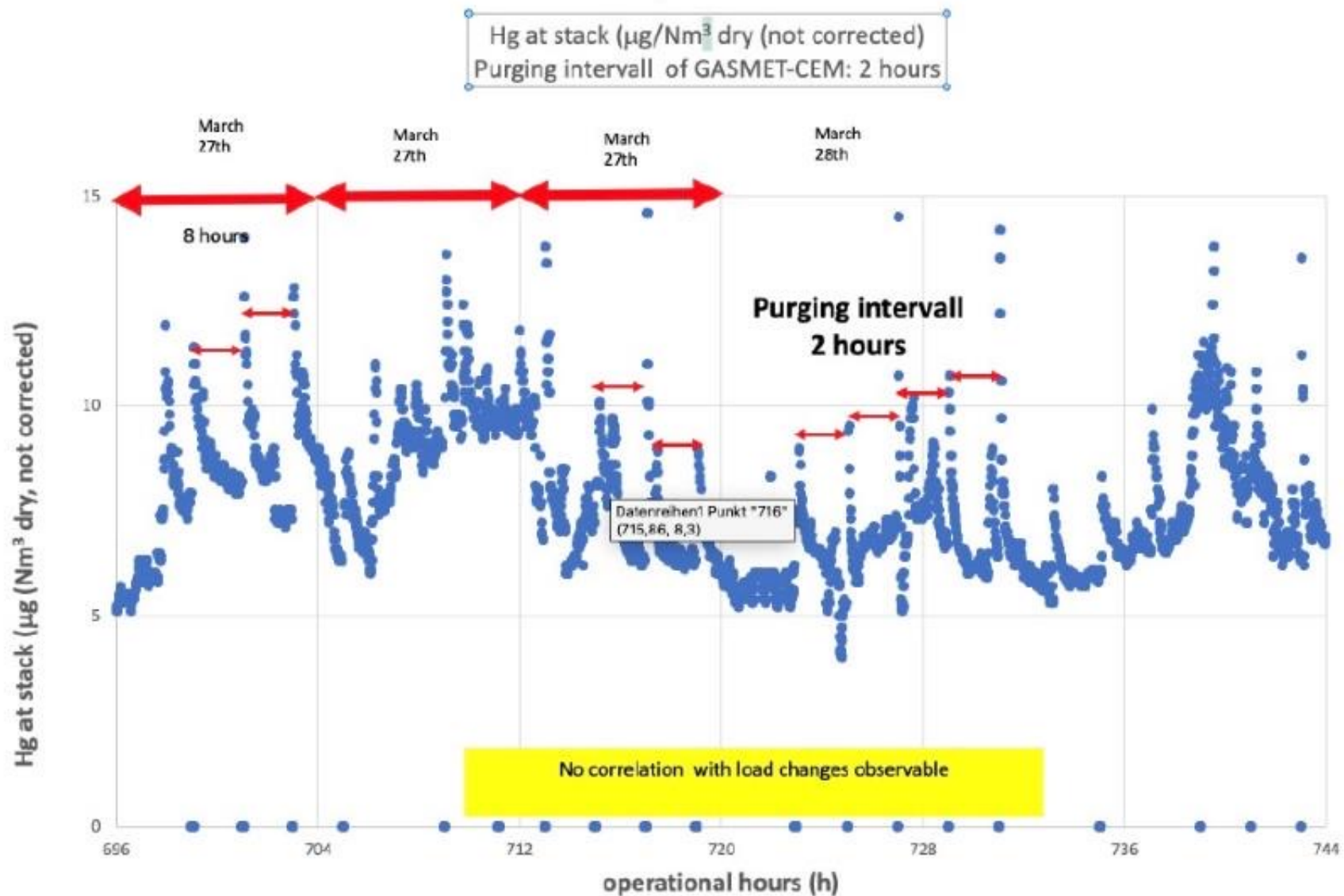
Both systems provide the highest sensitivity and annual availability on the market. CMM AutoQAL is the only TÜV and MCERTS certified solution with automatic and truly integrated QAL3 validation tool. Systems offer certified measurement with the lowest certified range in the world (0-5 $\mu\text{g}/\text{m}^3$). The highest certified measurement range of the system is 1000 $\mu\text{g}/\text{m}^3$ and even higher concentration peaks can be measured without any hardware changes. CMM has both TÜV and MCERTS (QAL1) certificates.

CMM AutoQAL and CMM both consist of:

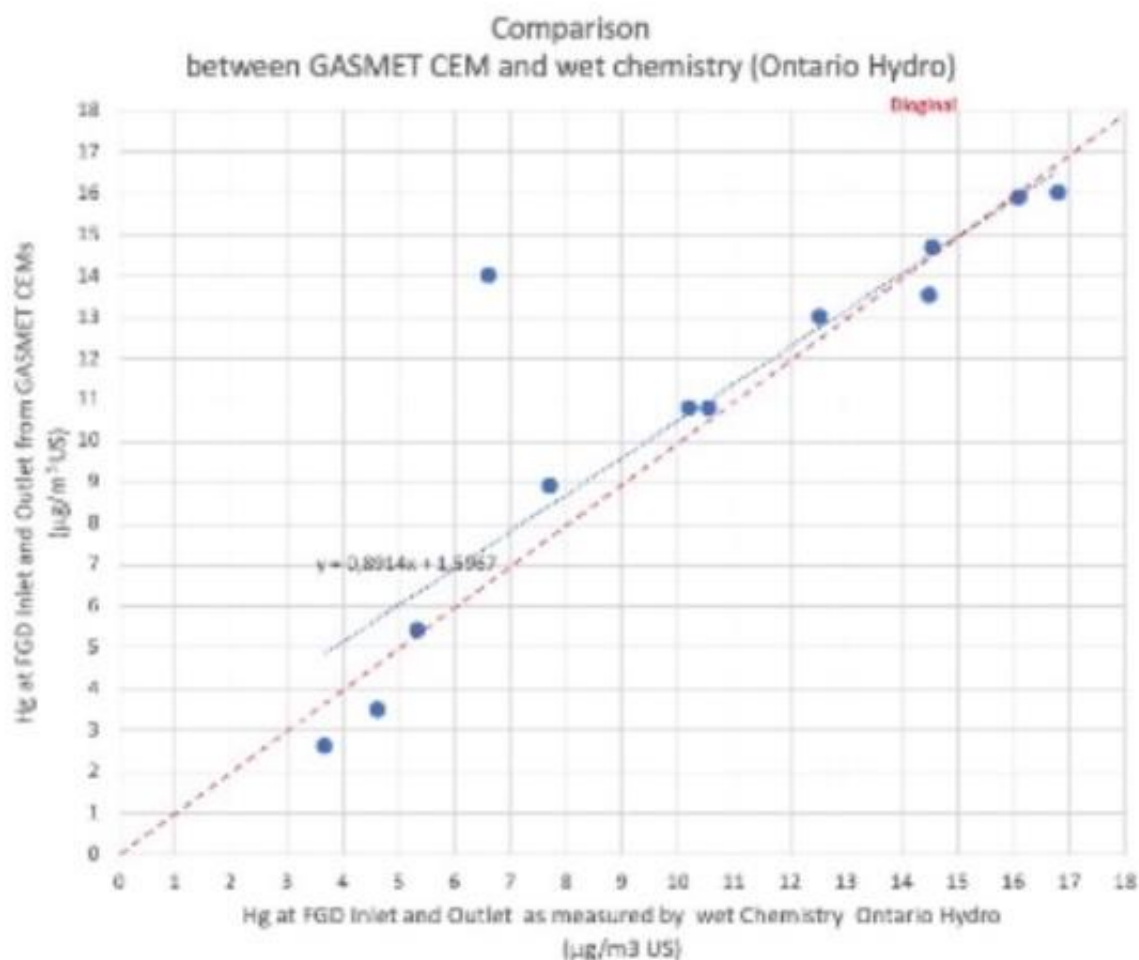
- Dilution probe
- Heated sample line
- Mercury analyzer
- Test gas generator

Gasmet mercury analyzer is based on Cold Vapor Atomic Fluorescence (CVAF) measurement principle, which gives the highest sensitivity in the world. The system is fully automatic, and the automatic calibrations are done by user-defined intervals. CMM AutoQAL has an integrated and certified test gas generator with possibility to do both Hg_0 and HgCl_2 checks means that there is no longer a need for an external gas generator for QAL3 operations.





Control of the GASMET CEM measurements by comparison with wet chemistry (Ontario Hydro) - should be extended especially for the Hg-levels $< 5 \mu\text{g}/\text{m}^3$



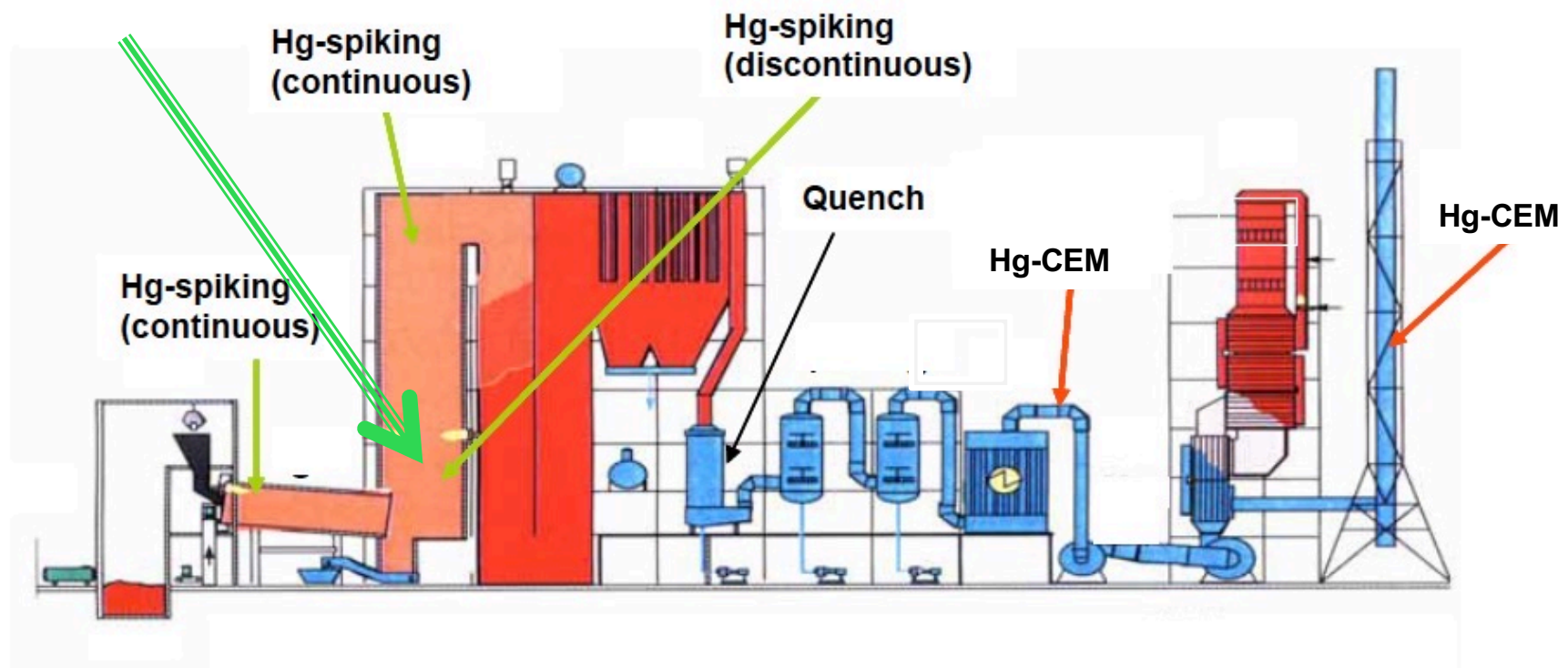
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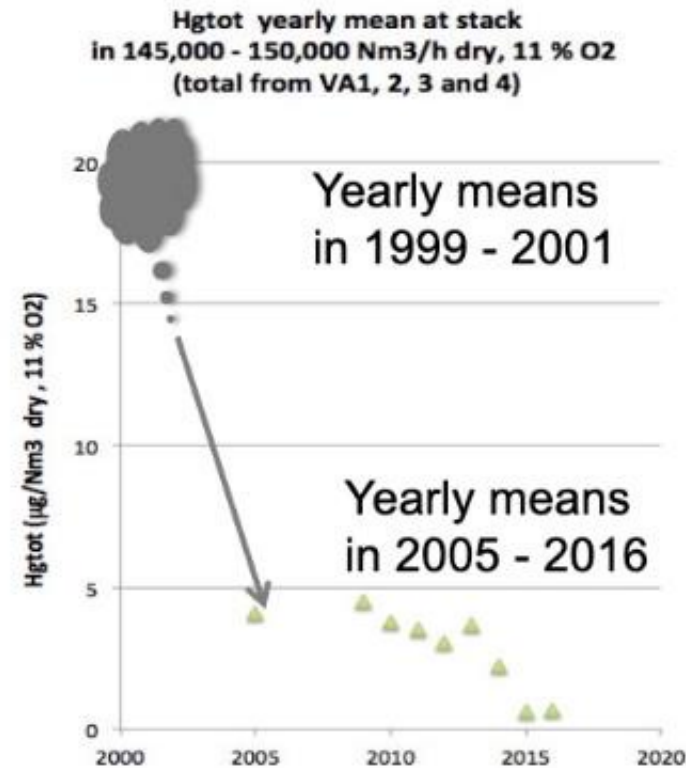
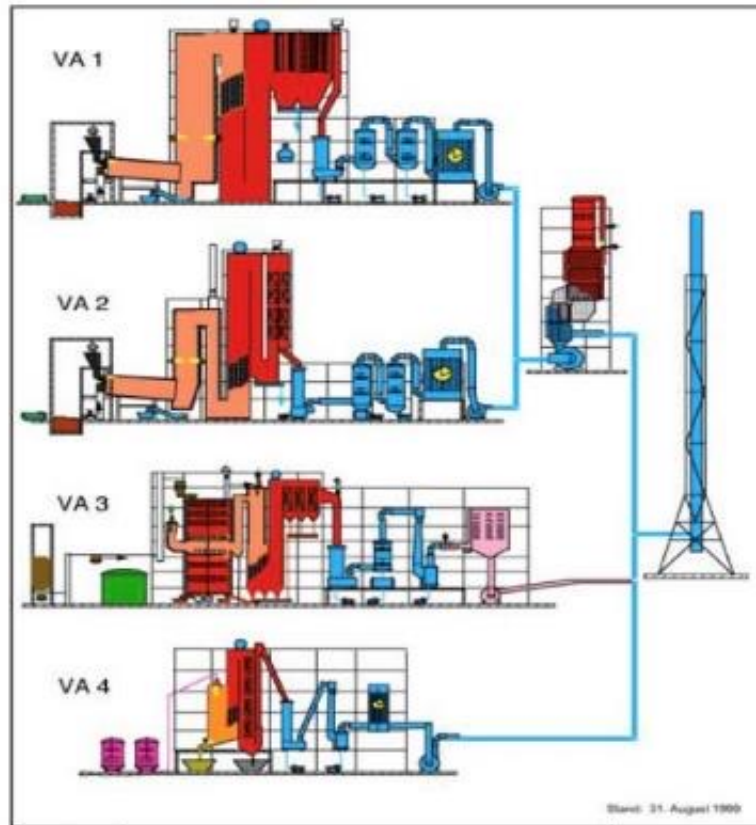
Bromides applied as Mercury Oxidizers by BAYER/CURRENTA since 2000

injecting bromide solutions (HBr , NaBr , CaBr_2)

Bromides

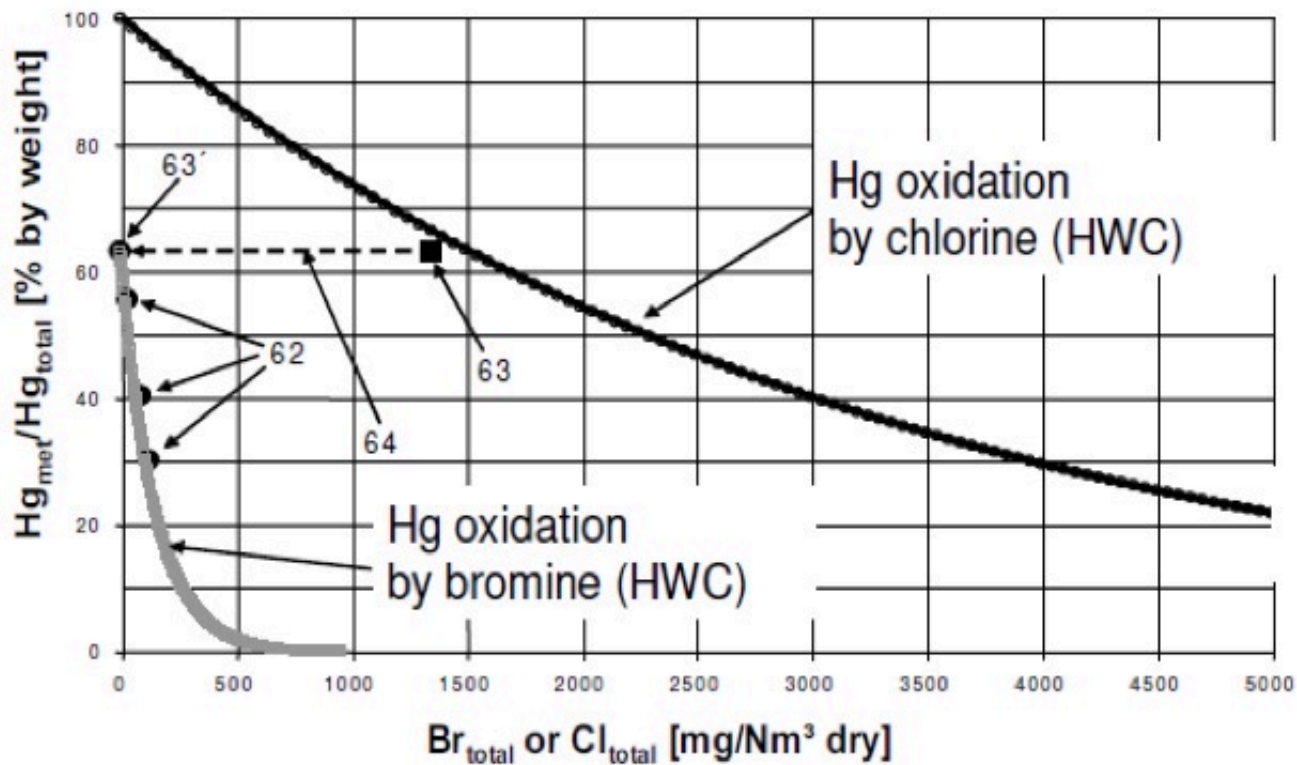


Invention Prof. Vosteen (2000)



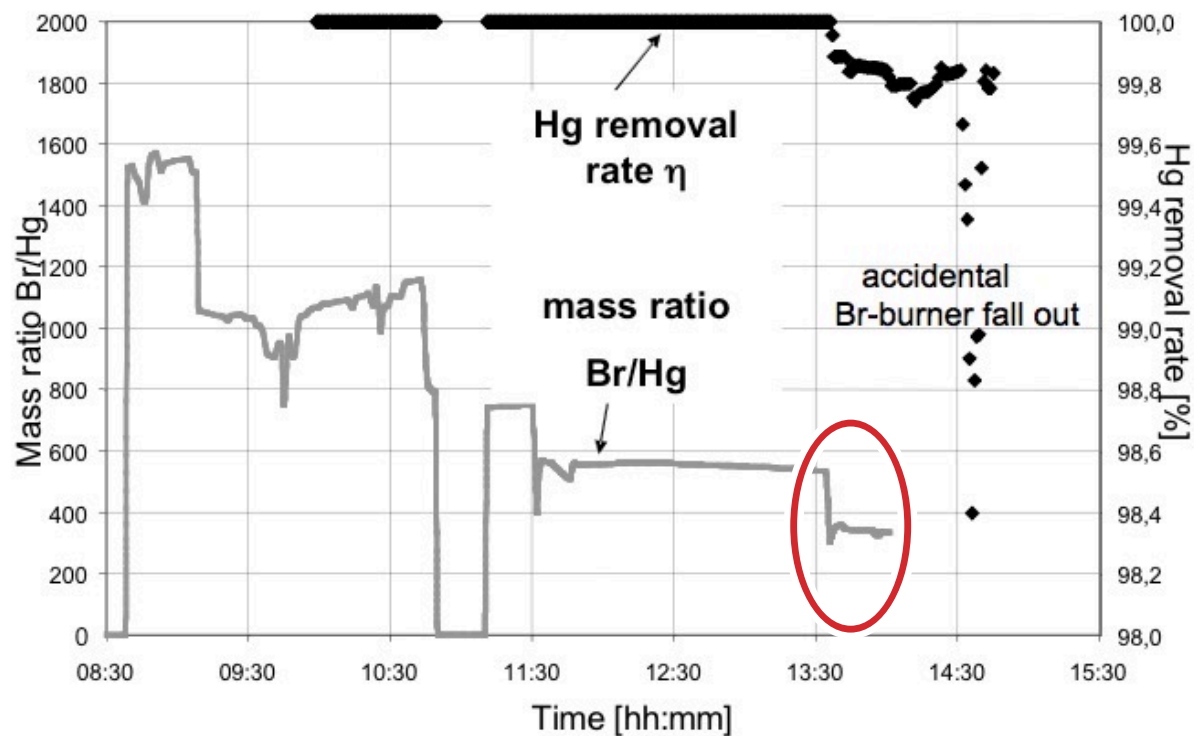
**Mercury concentration Hg_{tot} at the stack of the
CURRENTA Waste Management Center in Leverkusen-Bürrig**

Example: Bromine versus Chlorine



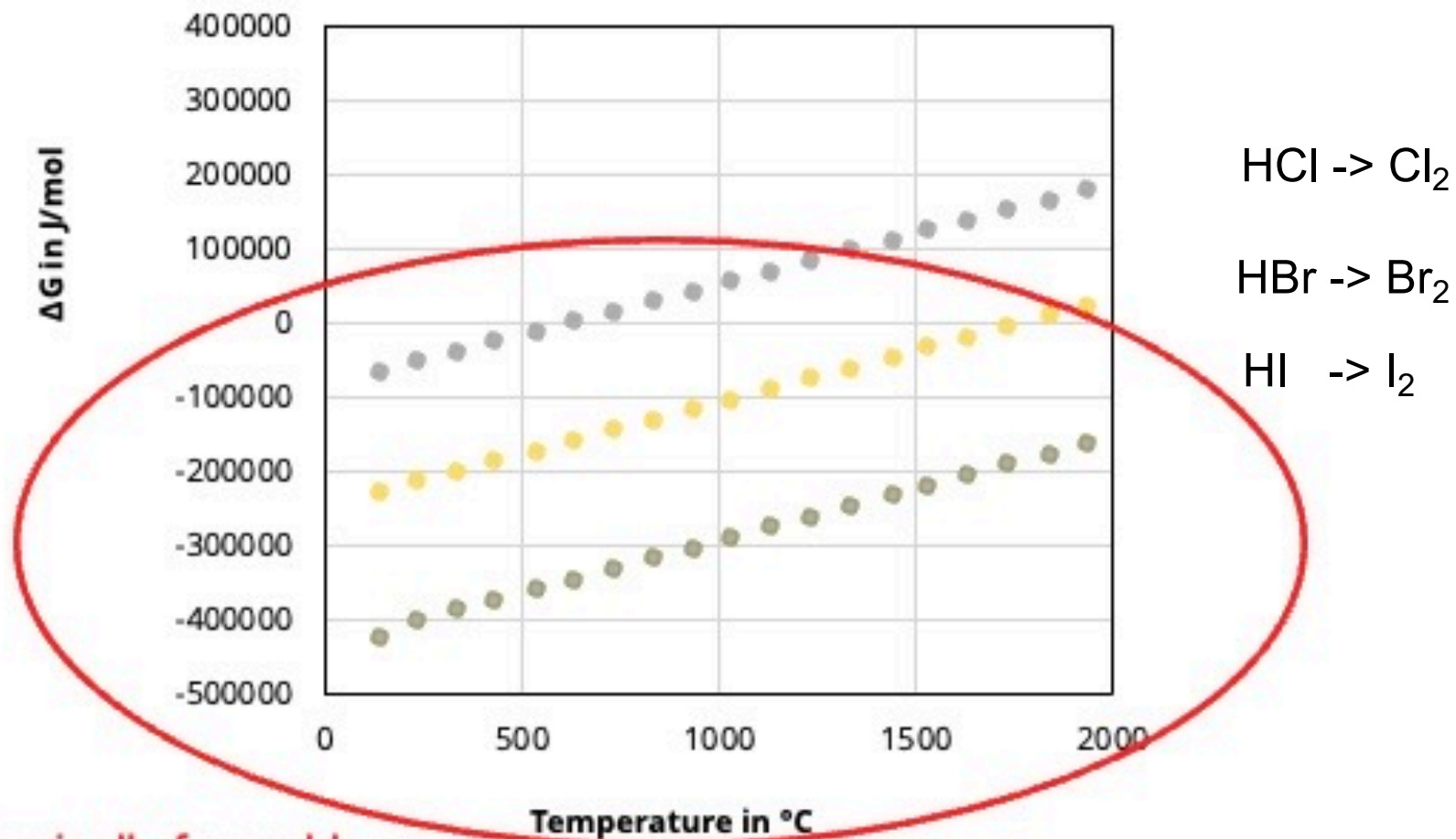
Source: B. W. Vosteen et al., EP 1 386 655 B1

Spiking the boiler raw gas with 9600 $\mu\text{g Hg/Nm}^3$ dry)



**Mass ratio Br/Hg = 100 ... 500 needed
("without high dust SCR")**

Deacon Equilibria for X = Br, Cl and I
 $4\text{HX} + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 2\text{X}_2$



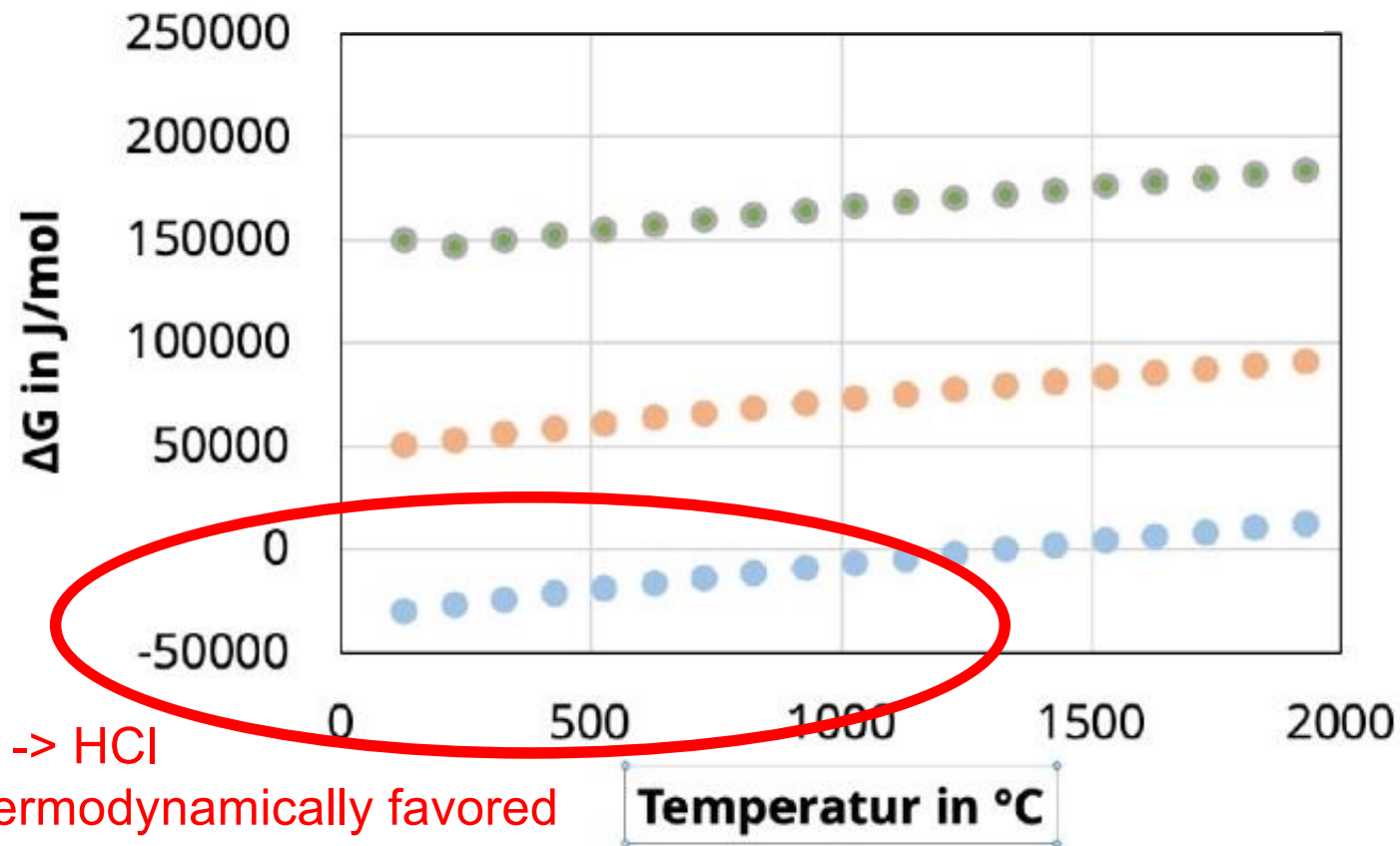
HCl \rightarrow Cl₂

HBr \rightarrow Br₂

HI \rightarrow I₂

Thermodynamically favorable

Griffin Equilibria for X = Br, Cl and I
 $\text{SO}_2 + \text{H}_2\text{O} + \text{X}_2 \rightarrow \text{SO}_3 + 2\text{HX}$



I₂ -> HI

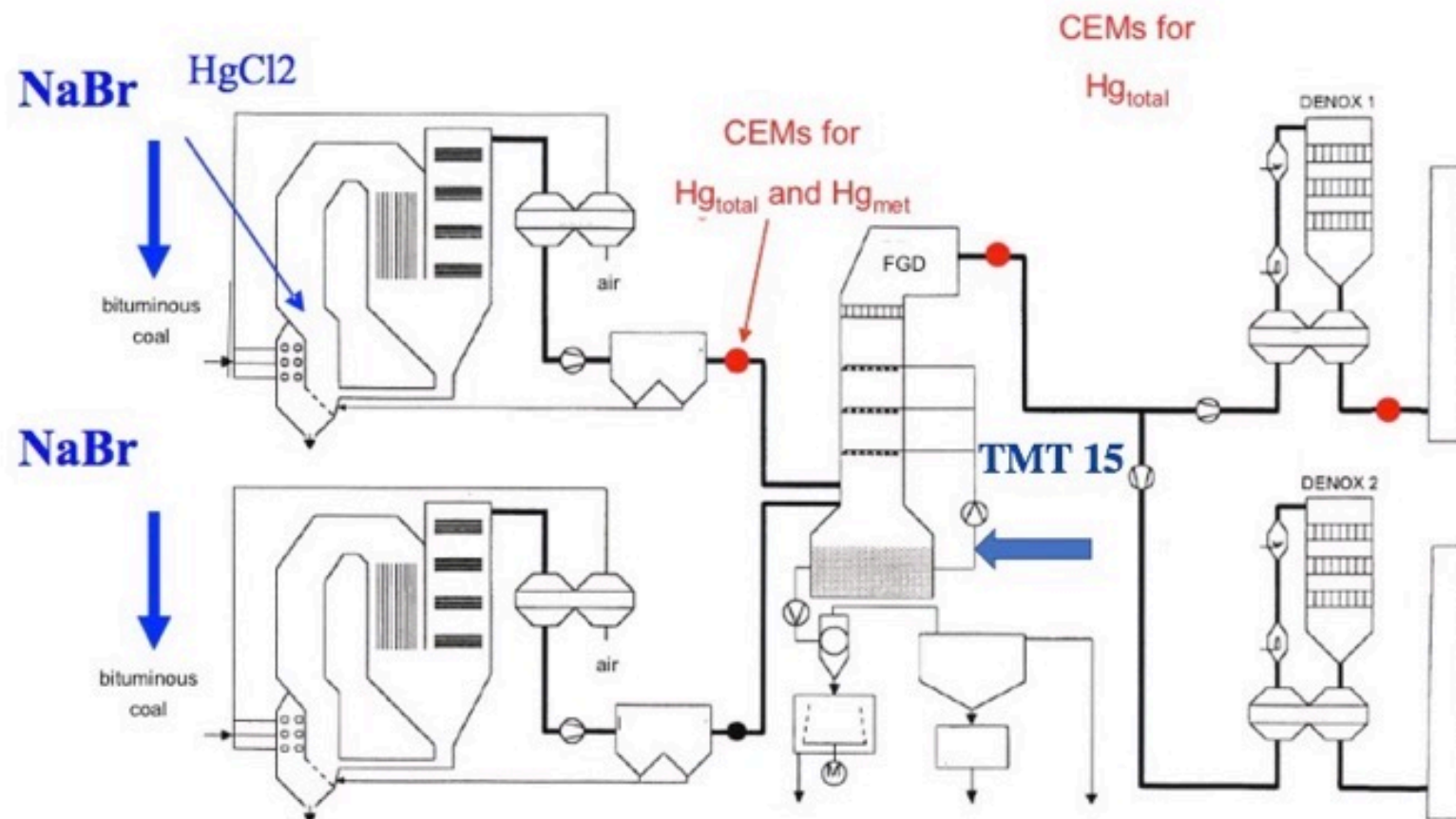
Br₂ -> HBr

Cl₂ -> HCl

Cl₂ -> HCl

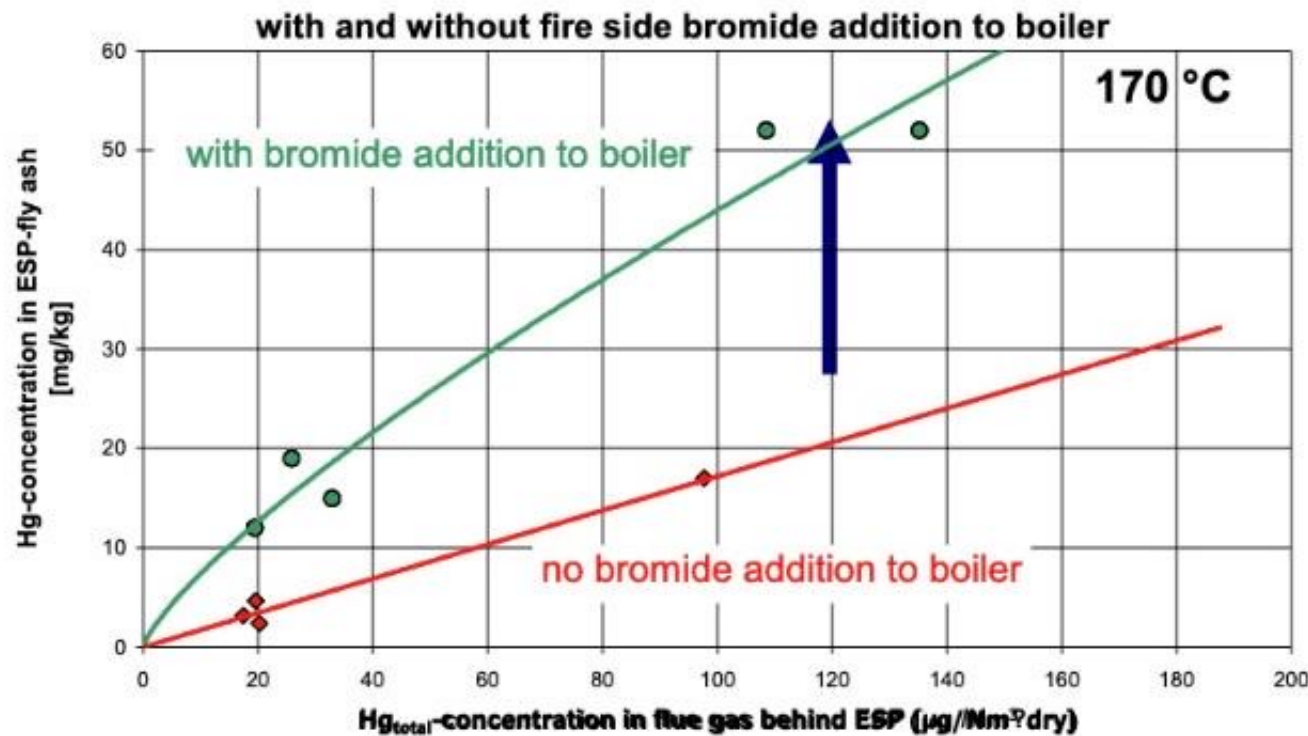
Thermodynamically favored

**Industrial Tests at coal-fired utility of BAYER/CURRENTA
with NaBr and TMT15 in 2001
(Power Plant N230 in Uerdingen)**



Fire side bromide addition (internal bromination) is improving Hg-adsorption at fly ash UBC by the factor 2,5

Sorption isothermes of the ESP-fly ash of high LOI (22,3 % UBC)



Source: Vosteen, Nolte et al.: „Chlorine- and bromine-enhanced Hg sorption on ESP flyash from a coal-fired slag-tap boiler and on cement raw meal“, VDI Seminar 431802, September 29 - 30, 2003 in Duesseldorf (Germany)

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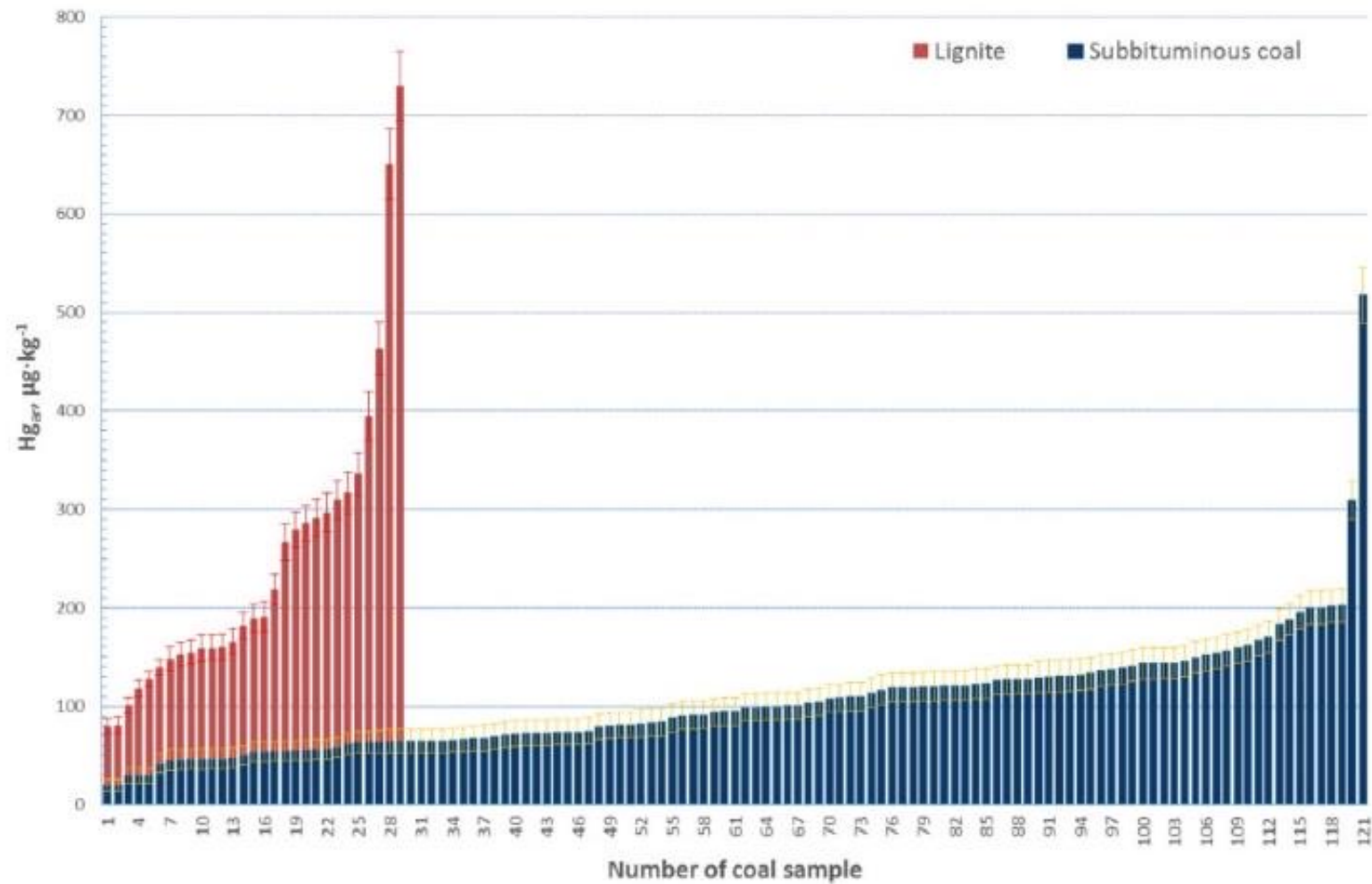


Fig. 3. Mercury content in analyzed samples.

Table 3

Proximate and ultimate analysis of tested coals.

Component	Plant 1	Plant 2	Plant 3
Hg (ppb)	66 ± 9	100 ± 15	596 ± 99
Cl (ppm)	1040 ± 50	750 ± 45	88 ± 12
Br (ppm)	15.0 ± 1.1	14.1 ± 0.6	4.2 ± 0.1
S (wt.%)	0.49 ± 0.01	0.64 ± 0.04	1.18 ± 0.06
Moisture ^a (wt.%)	9.0 ± 0.4	9.8 ± 0.7	52.9 ± 0.6
Ash (wt.%)	26.8 ± 1.6	20.5 ± 2.4	25.3 ± 2.9
q _{p,net} ^a (MJ kg ⁻¹)	22.91 ± 0.65	23.14 ± 0.28	7.42 ± 0.37

^a Dry basis or as-received.

corresponds to Bełchatów Lignite as fired

Mercury Balance performed at Bełchatów Unit 4 in 2016

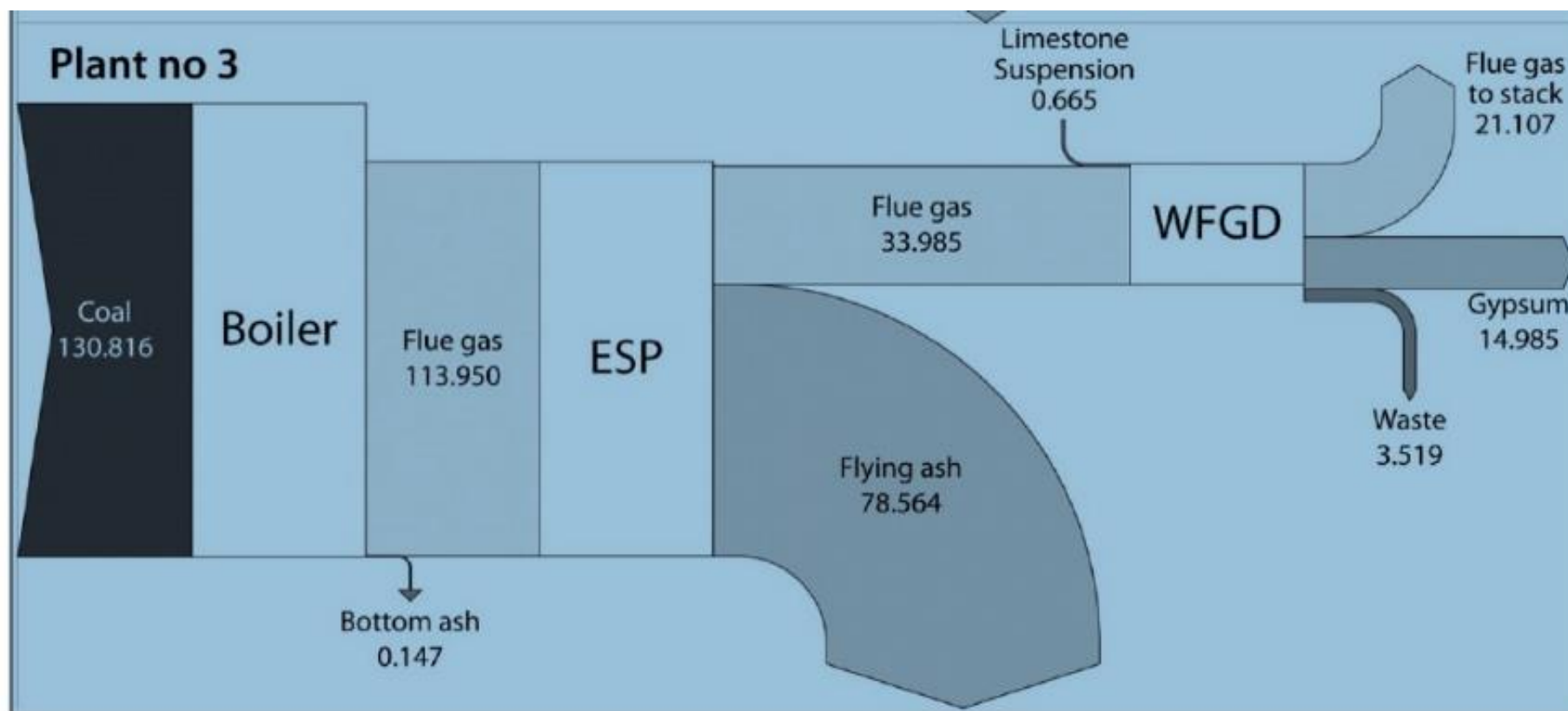
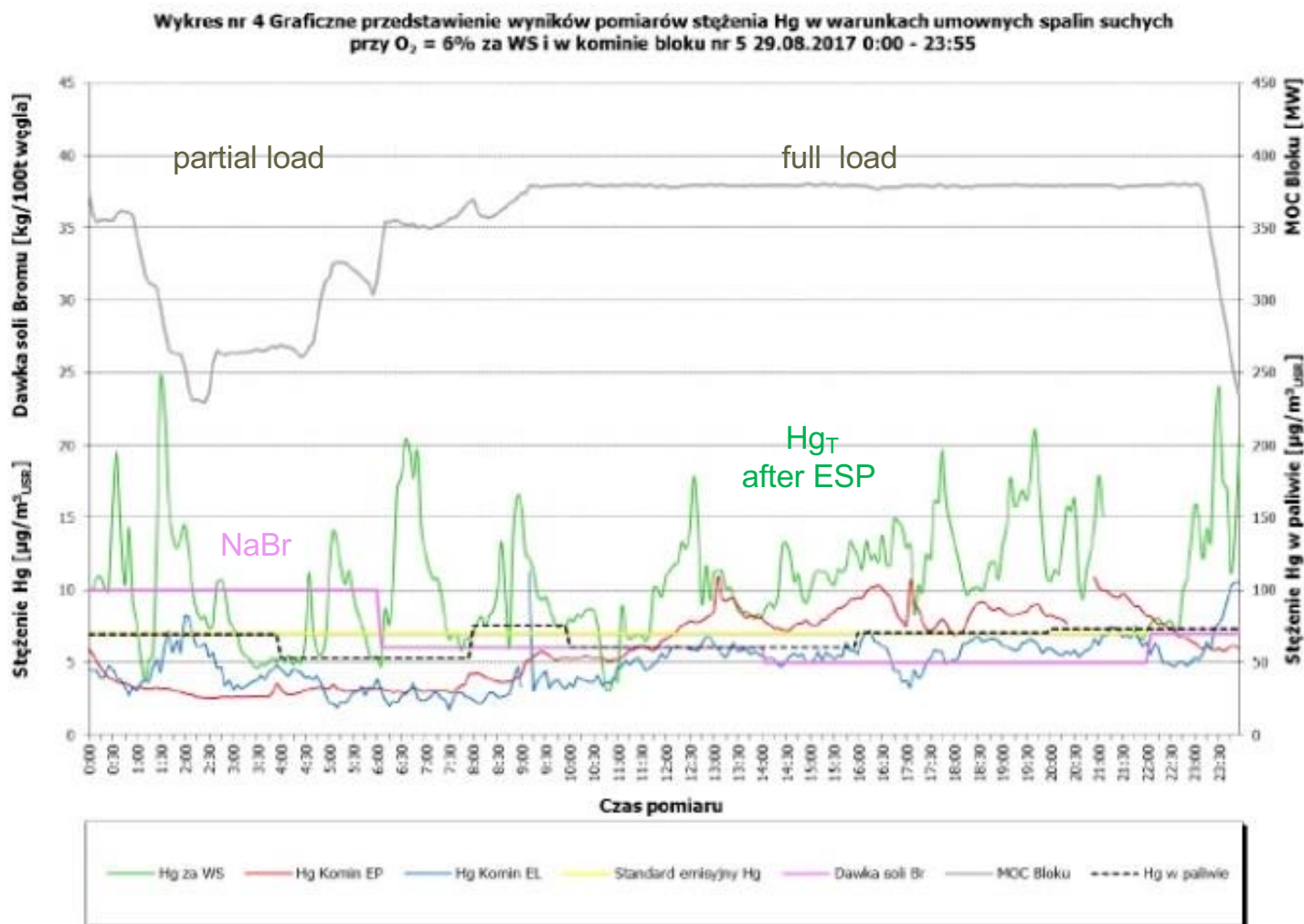


Fig. 7. Sankey diagrams of mercury distribution (mercury stream, g·h⁻¹).

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Promissing test results at Unit 5 in August 29th, 2017

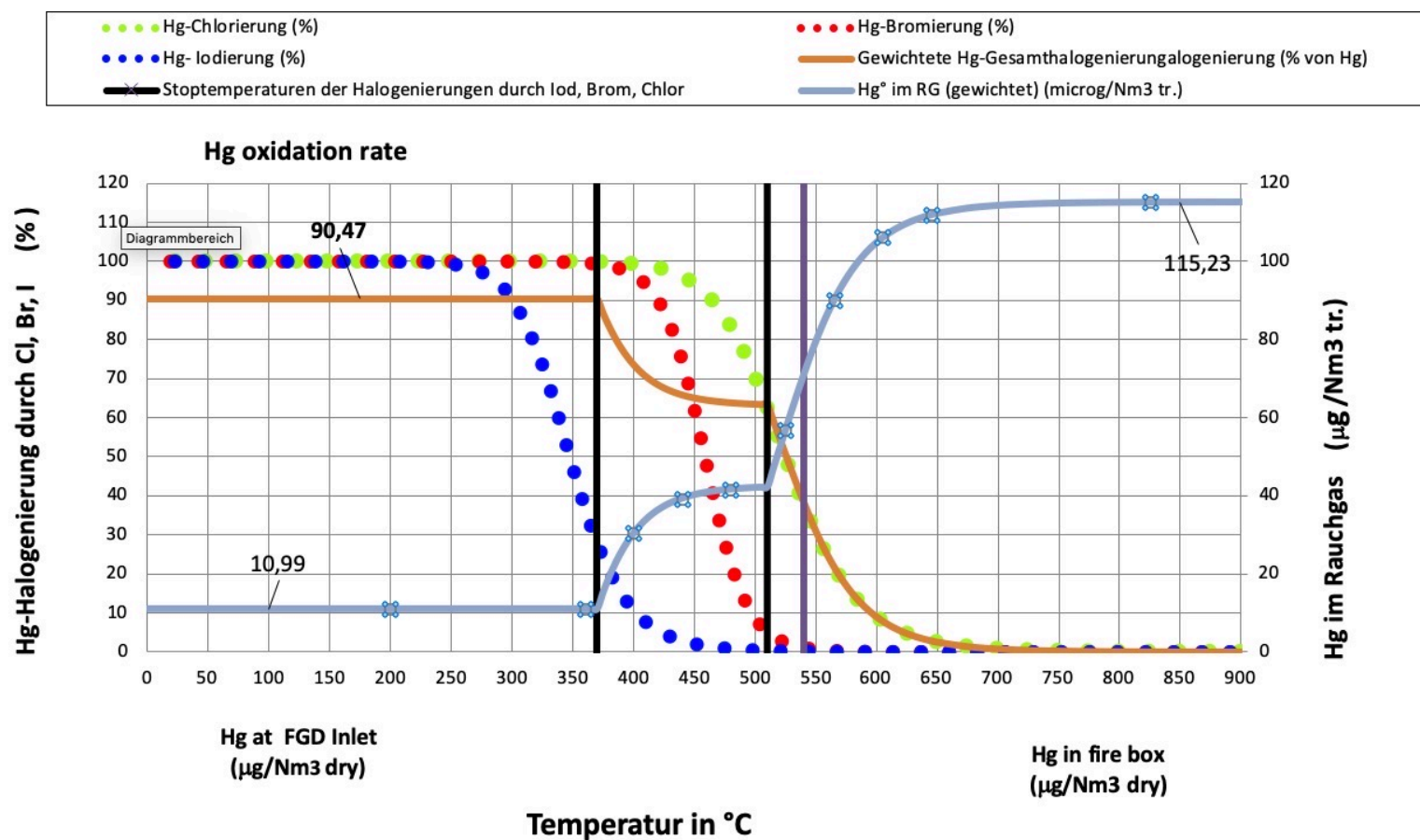


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Modelling of the mercury oxidation by halogens (Stoptemperature method of Prof. Vosteen)

Example for a calculation point
with given Cl (native) , Br (not added) and I (added)



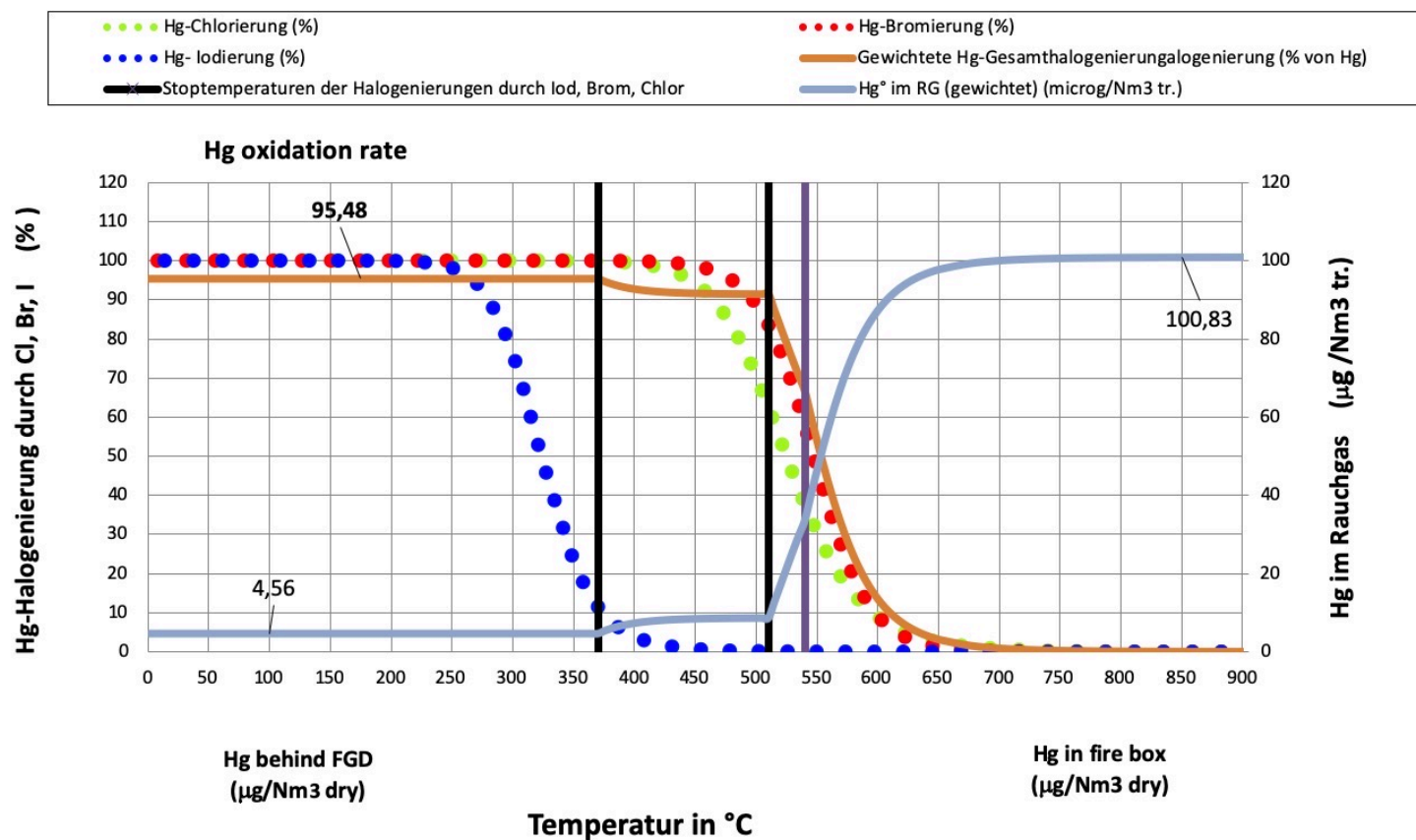
Analyzing the Test Results from 2017/2018 with help of the Vosteen's modelling programm delivered the following Stoptemperatures:

Stopptemperatur der Hg-Chlorierung(°C)	510
Stopptemperatur der Hg-Bromierung (°C)	540
Stopptemperatur der Hg-Iodierung (°C)	370

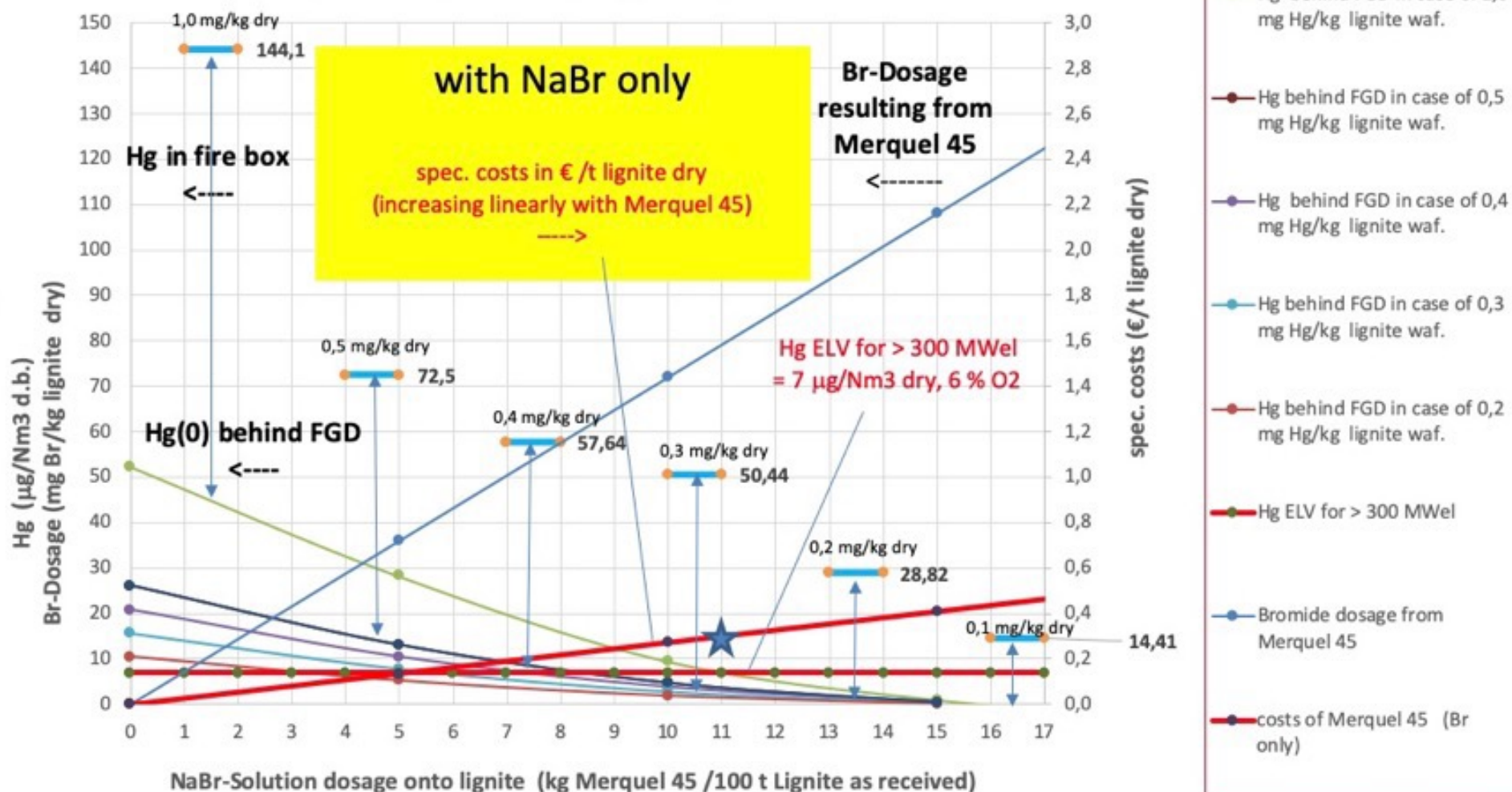
Fig. 14: Defined Stop-Temperatures chosen for Unit 3^{*)}

^{*)} Please note: The model was written in German language (engl. Stop-Temperature is Stopptemperatur in German).

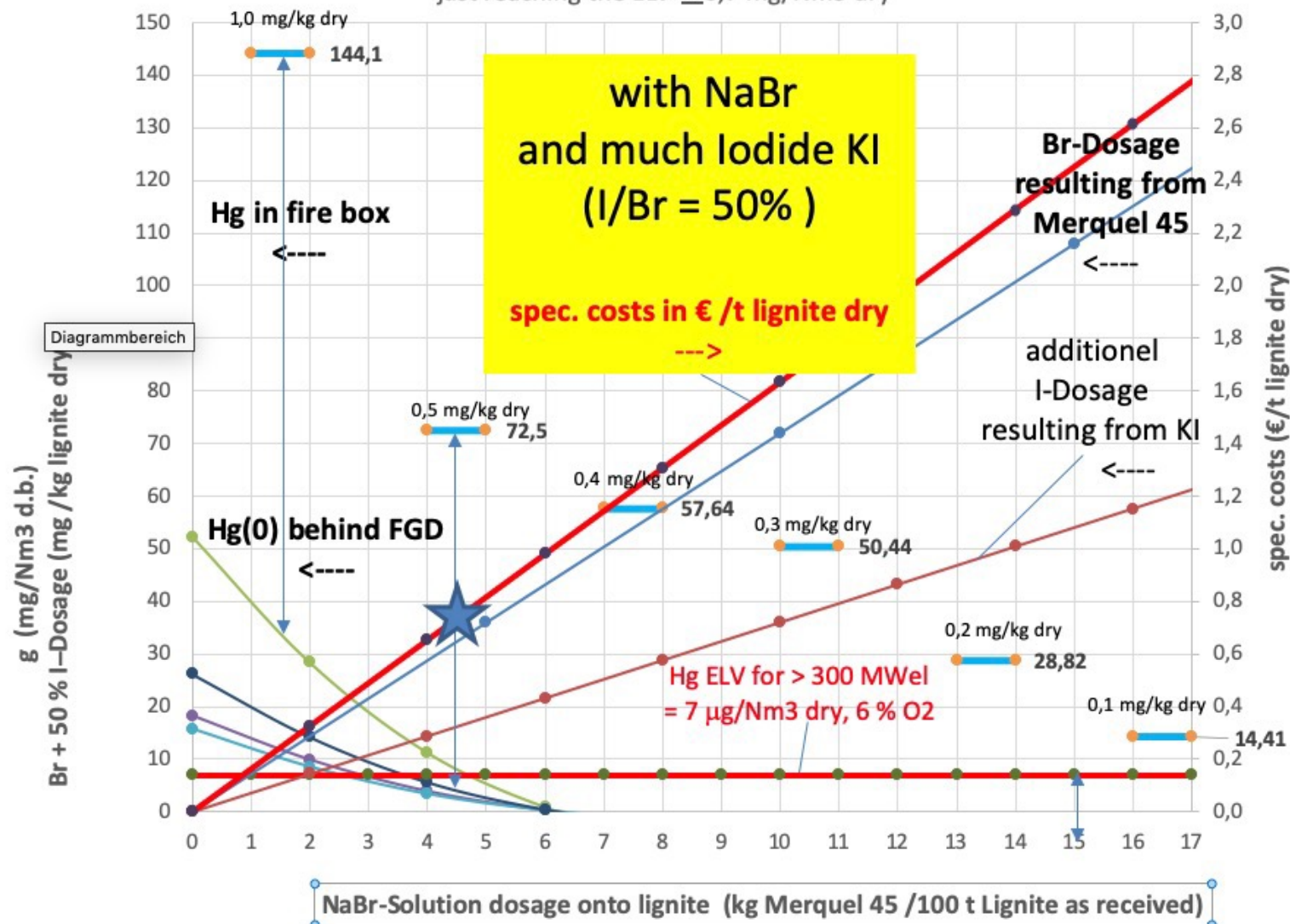
Example for a calculation point
with given Cl (native) , Br (added) and I (added)



Working Range at Unit 3 and Unit 14 -
preliminary calculation of the specific halide costs -
Blue star indicates the value for
1 mg Hg/kg lignite dry
just reaching the ELV $\leq 0,7 \text{ mg/Nm}^3 \text{ dry}$



Working Range at Unit 3 and Unit 14 -
preliminary calculation of the specific halide costs -
Blue star indicates the value for 1 mg Hg/kg lignite dry
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Long-Time Testings at Bełchatów Unit 3 in 2021

Belchatów Unit 3, tests March 14th- March 28, 2021

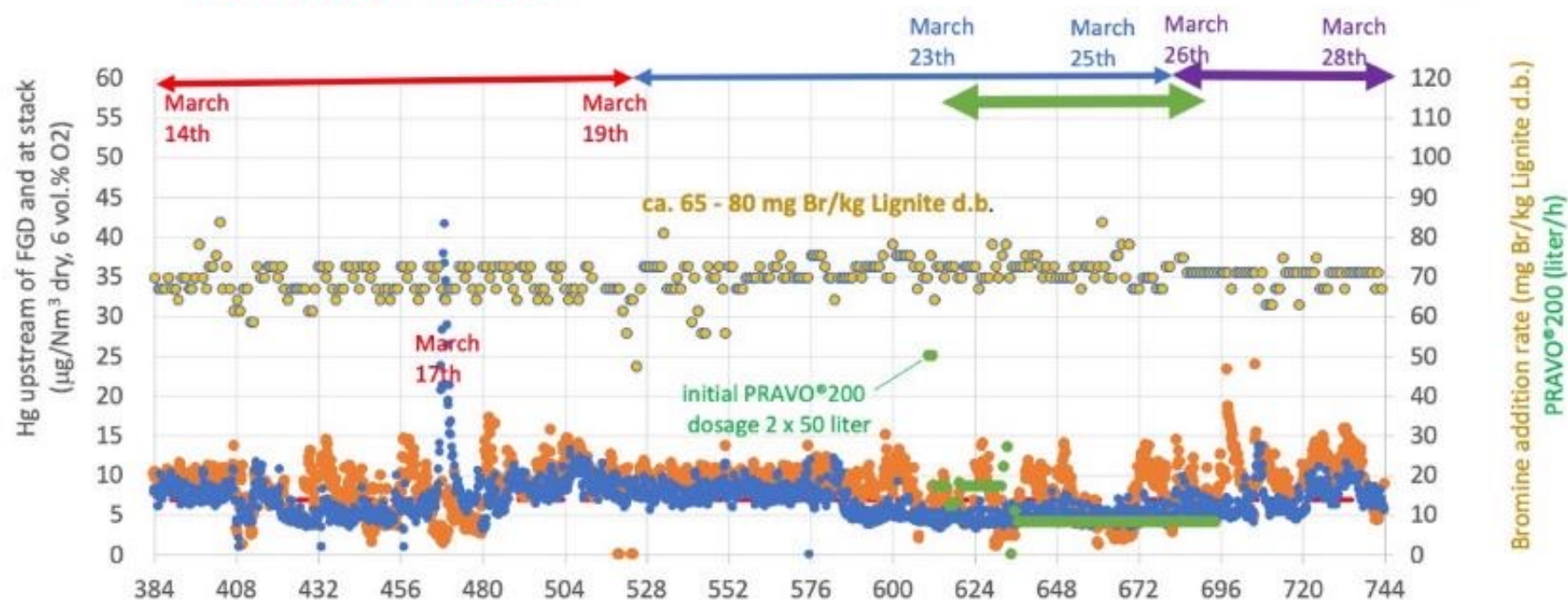
Hg upstream FGD and downstream FGD

• WLOT $\mu\text{g}/\text{m}^3\text{US}$ • Hg at stack $\mu\text{g}/\text{m}^3$ with O₂-correction — ELV limit • PRAVO addition to FGD

with NaBr + ADD

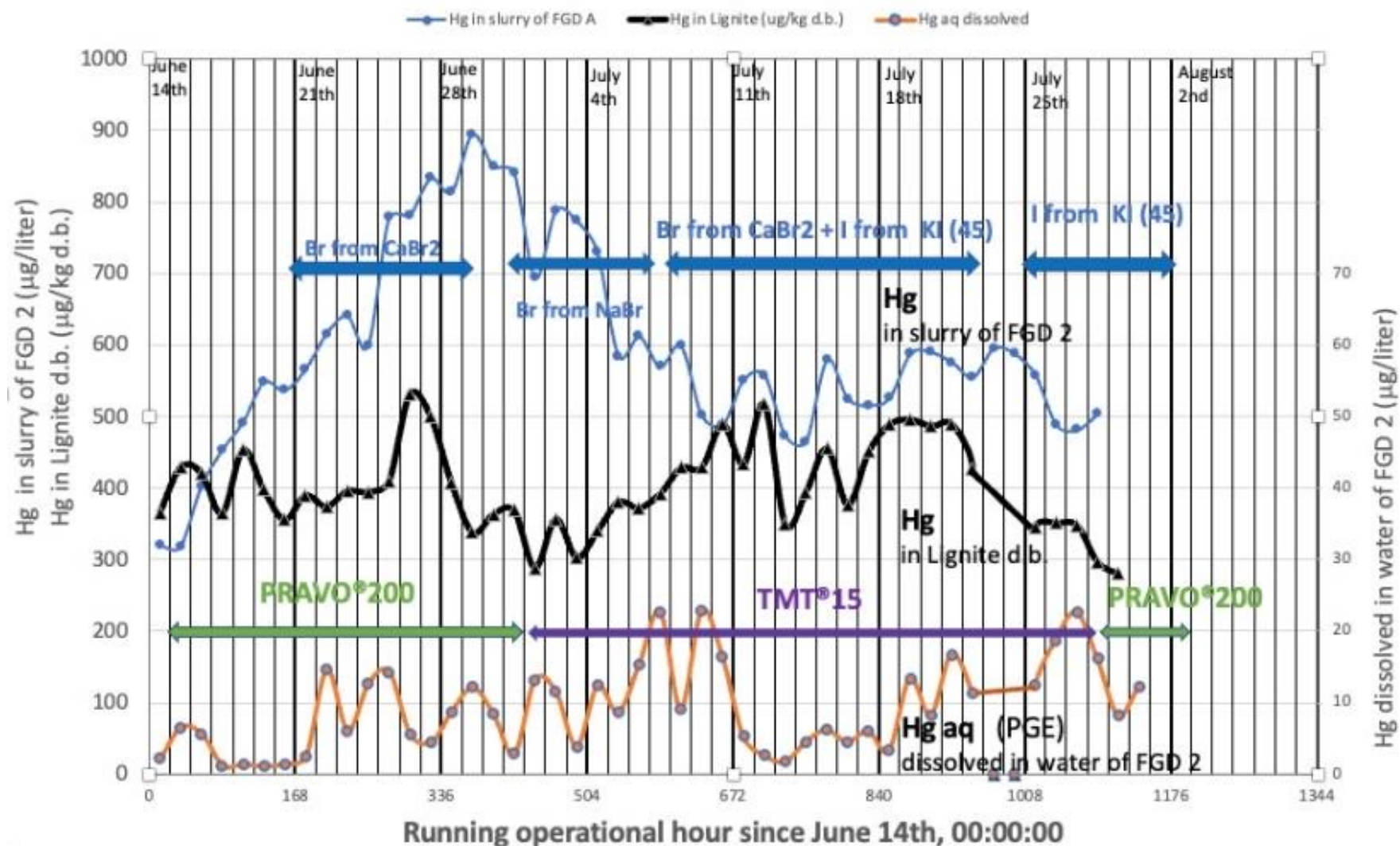
with NaBr and PRAVO®200

with CaBr₂



Long-Time Testings at Bełchatów Unit 14 in 2021

Unit 14 : Hg accumulation in FGD 2

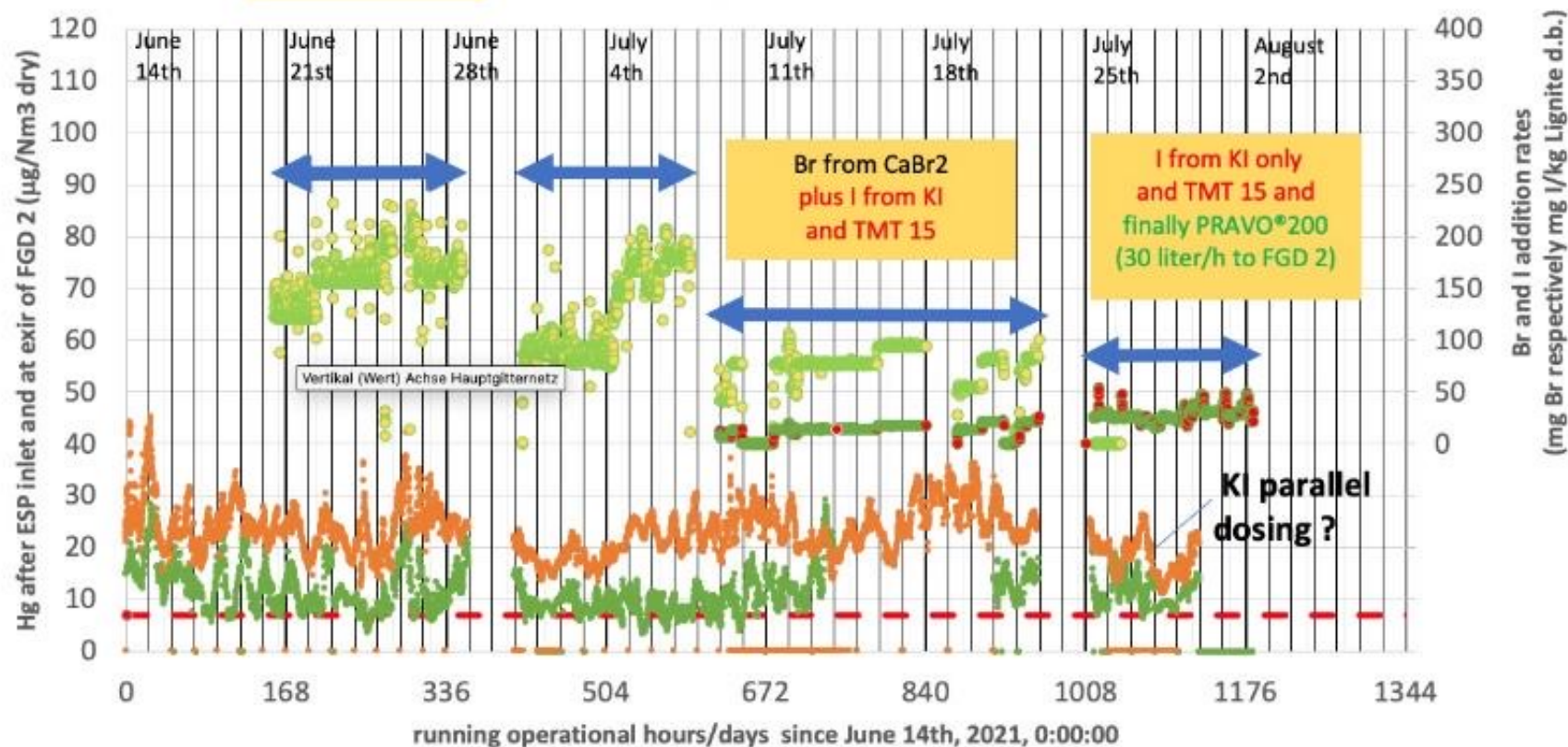


Tests at Unit 14 (858 MWel)

- ELV Emission Limit Value
- Hg after ESP uncorrected
- I dosing rate (mg I/kg Lignite d.b.)
- Hg at FGD 2 exit (corrected to 6 vol.-% O₂)
- Br dosing rate (mg Br/kg Lignite d.b.)

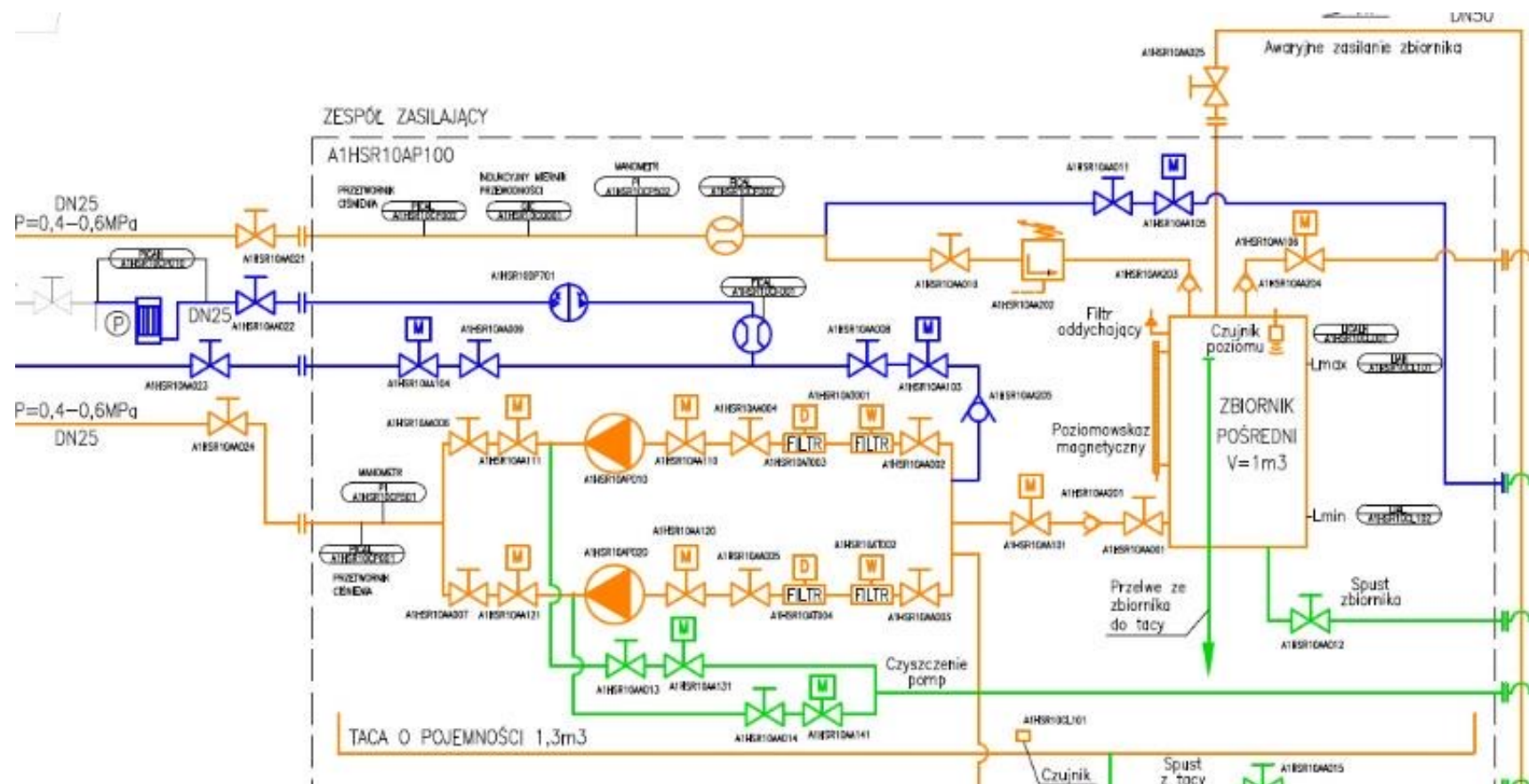
Br from CaBr₂
and PRAVO®200

Br from NaBr
and PRAVO®200

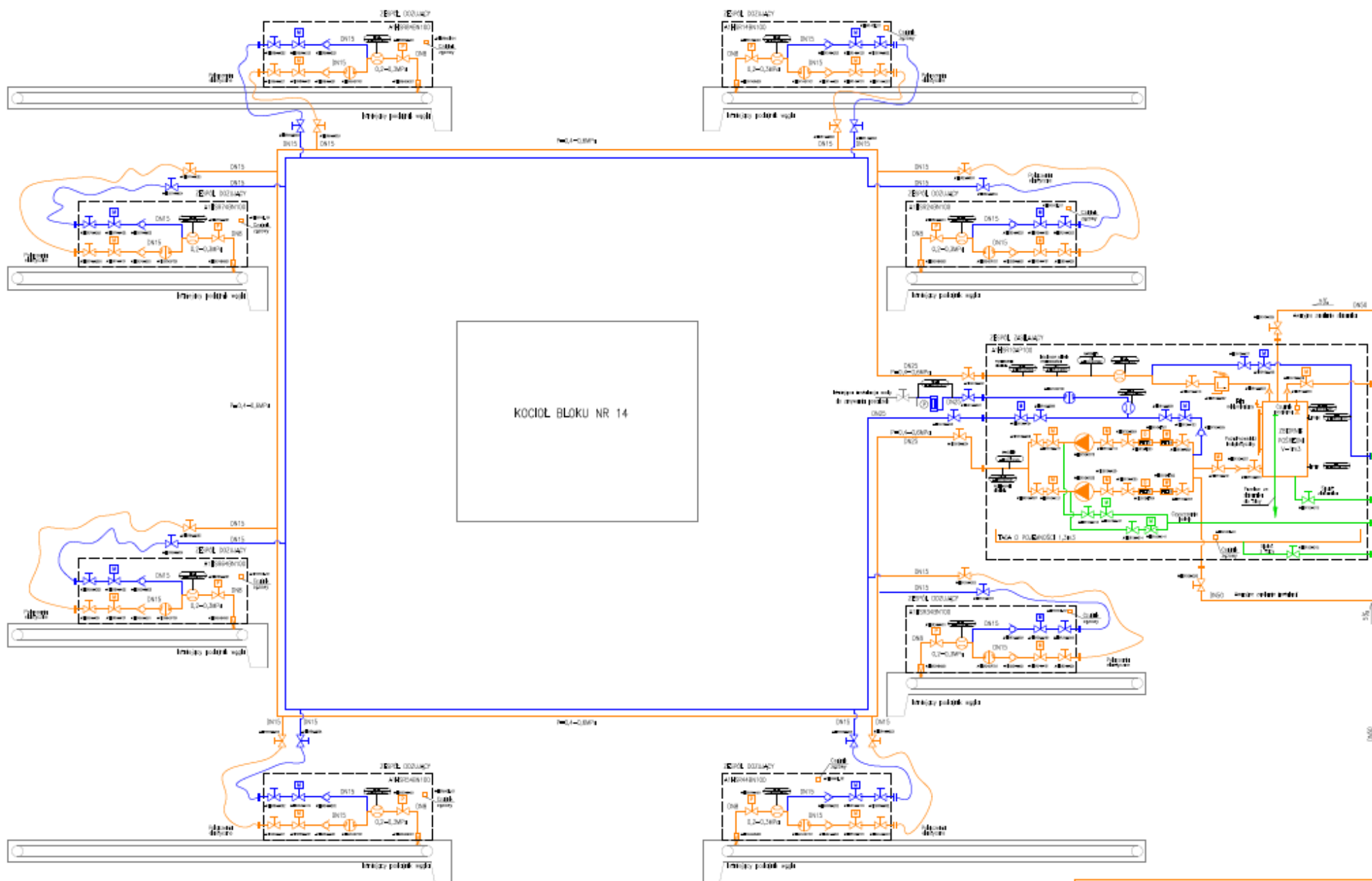


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Design of the Bromide Solution Feeding System



Overview of the Bromide Solution Feeding System



Storage tanks

2 x 60 m³

serving the Units 2-12





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Conclusion

Emission levels $<7 \mu\text{g}/\text{Nm}^3$ d.b. at stack can be achieved with limited dosing rates of NaBr or CaBr_2 solutions, as well.

NaBr is getting more and more expensive. Therefor CaBr_2 is recommended.

Iodides are far more effective in mercury oxidation than bromides, iodide salts or solutions might therefor be used, as well - e.g. in form of KI or NaI-solutions But such iodides have got extremely expensive. Not to be recommended.

Precipitation agents as anorganic PRAVO®200 or organic TMT®15 can both be used as FGD additive to suppress Hg-reemissions and to stabilize the dissolved mercury as solid mercury sulfide or mercury-TMT-complex.

During full load, the oxygen content should not be lowered too much. Otherwise the mercury oxidation will become insufficient (Deacon reaction needs oxygen).